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EXPERIMENTAL SET-UP

Samples: standard factory product cocoa butter and cocoa butter from Nigeria

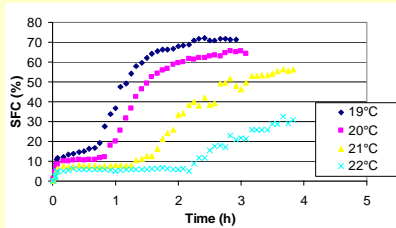
Analytical techniques: DSC: sample preparation, 65°C / 15 min (removal of memory effect), cooling at 8°C / min to crystallization temperature, hold until crystallization is finished

pNMR: sample preparation, 65°C / 1h (removal of memory effect), transfer to cryostat at crystallization temperature, take readings at appropriate intervals

Crystallization temperatures: between 19°C and 23°C (0.5°C interval for DSC experiments, 1°C interval for pNMR experiments)

Analyses: step 2 of the DSC curves is integrated and modeled by the Avrami and Gompertz models

INFLUENCE OF TEMPERATURE AS MEASURED BY pNMR



INFLUENCE OF TEMPERATURE ON STEP 2 OF THE CRYSTALLIZATION PROCESS AS MEASURED BY DSC

AVRAMI MODEL

$$f(t) = a * (1 - e^{-k * t^n})$$

$f(t)$: amount of crystallization at time t [J/g]

a : value for f as t approaches infinity [J/g]

k : crystallization rate constant [h^{-n}]

n : Avrami exponent [-]

GOMPERTZ MODEL

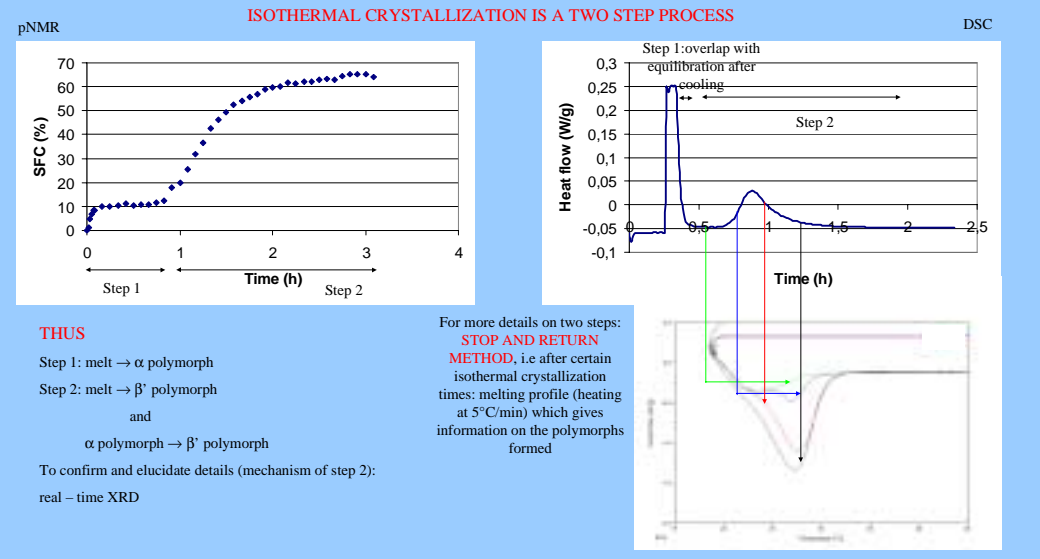
$$f(t) = a * e^{-e^{-\frac{\mu}{\lambda} * (\lambda - t) + 1}}$$

$f(t)$: amount of crystallization at time t [J/g]

a : value for f as t approaches infinity [J/g]

μ : maximum increase rate [J/g h^{-1}]

λ : induction time [h]



CONCLUSIONS

- Isothermal crystallization of cocoa butter is a two-step process, starting with some crystallization in the α polymorph, followed by (re)crystallization in the β' polymorph
- The amount of α crystals formed decreases as temperature increases
- The final SFC value decreases as temperature increases, while the a parameter of the second step as measured by DSC only starts to decrease at high temperatures
- The induction time increases as temperature increases, while the maximum increase rate and the Avrami rate constant decrease as temperature increases
- There is no clear dependence of the Avrami exponent on the temperature

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