

Integrated water quality modeling:  
New perspectives using IWA  
RWQM1 model

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Universiti Teknologi Malaysia, November 17 2000

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**IWA RIVER WATER QUALITY MODELLING  
(RWQM) TASK GROUP**

L. Somlyódy, D. Borchardt, M. Henze,  
W. Rauch, P. Reichert, P. Shanahan, P. Vanrolleghem

**INTRODUCTION AND BACKGROUND**

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**SHORTCOMINGS IN RWQM  
FORMULATIONS**

- **LACK OF CLOSED MASS BALANCES:**  
BOD AND SEDIMENT
- **LACK OF SEDIMENT RELATED PROCESSES**  
(attached bacteria/algae and other benthic flux terms)
- **INCONSISTENCY IN RWQM  
AND ASM FORMULATIONS:** lack of integrating  
wastewater treatment and river water quality

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### SHORTCOMINGS IN RWQM FORMULATIONS

- SEWER OVERFLOW PROBLEMS
- SPATIAL NON-UNIFORMITIES AND RAPID TEMPORAL CHANGES
- ISSUES OF CALIBRATION
- PREDICTIVE CAPABILITY

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### OBJECTIVES OF THE RWQM TASK GROUP

- ☞ SCIENTIFIC AND TECHNICAL BASE
- ☞ “STANDARDIZED” RWQM CONVERSION MODEL VERSIONS
- ☞ GUIDELINES FOR MODEL SELECTION
- ☞ CASE STUDIES
- ☞ OPEN ENDED DEVELOPMENT PROCESS

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### RIVER WATER QUALITY MODEL NO. 1: I. MODELLING APPROACH

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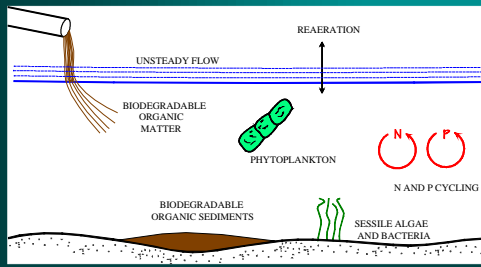
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## Modelling Context: River Water Quality




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## Mass Balance Equation

Change in concentration over time

$$\frac{\partial c}{\partial t} = \underbrace{-u \frac{\partial c}{\partial x} - v \frac{\partial c}{\partial y} - w \frac{\partial c}{\partial z}}_{\text{Advection}} + \underbrace{\frac{\partial}{\partial x} \left( \epsilon_x \frac{\partial c}{\partial x} \right) + \frac{\partial}{\partial y} \left( \epsilon_y \frac{\partial c}{\partial y} \right) + \frac{\partial}{\partial z} \left( \epsilon_z \frac{\partial c}{\partial z} \right)}_{\text{Dispersion}} + \underbrace{r(c, p)}_{\text{Reaction}}$$

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## Model Decision Procedure

- Define temporal representation
- Determine spatial dimensions
- Determine representation of mixing
- Determine representation of advection
- Determine reaction terms
- Determine boundary conditions

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### Step 1. Define Temporal Representation

- Define upper- and lower-bound time constants
- Define water-column and sediment process time constants

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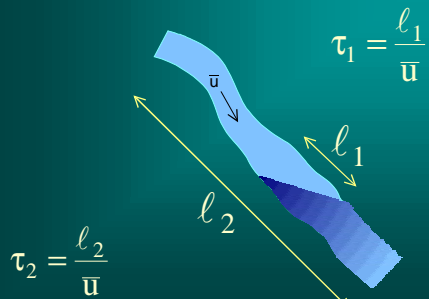
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### Upper- and lower-bound time constants



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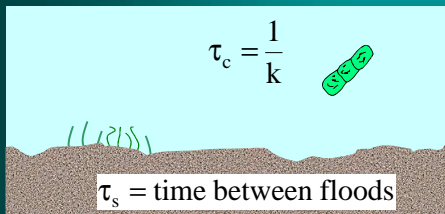
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### Water-column and sediment reaction time constants



Usually:  $\tau_s > \tau_c$

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### Step 1. Define Temporal Representation – Water-Column Reactions

$\tau_c \ll \tau_1 \rightarrow$  steady-state model  
 $\tau_1 < \tau_c < \tau_2 \rightarrow$  dynamic model  
 $\tau_c \gg \tau_2 \rightarrow$  reactions negligible



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### Summary: Model Decision Procedure

- Define temporal representation
- Determine spatial dimensions
- Determine representation of mixing
- Determine representation of advection
- Determine reaction terms
- Determine boundary conditions

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### RIVER WATER QUALITY MODEL NO. 1: II. BIOCHEMICAL PROCESS EQUATIONS

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## Contents

$$\frac{\partial c}{\partial t} = -u \frac{\partial c}{\partial x} - v \frac{\partial c}{\partial y} - w \frac{\partial c}{\partial z} + \frac{\partial}{\partial x} \left( \epsilon_x \frac{\partial c}{\partial x} \right) + \frac{\partial}{\partial y} \left( \epsilon_y \frac{\partial c}{\partial y} \right) + \frac{\partial}{\partial z} \left( \epsilon_z \frac{\partial c}{\partial z} \right) + r(c, p)$$

- Features
- Simplifying Assumptions
- Components
- Processes

**RWQM1 Peterson Matrix**

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## Features

- Compilation of formulations used in the literature with special emphasis on
  - compatibility with activated sludge models
  - closing of mass balances
  - calculation of pH

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## Simplifying Assumptions

- Constant elemental composition of compounds and of process stoichiometry.
- No adaptation.
- No anaerobic conditions (nitrate available).
- Influence of macrophytes as varying surface for sessile organisms neglected.
- No silicate limitation.

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## Components

Organisms:  $X_H, X_{N1}, X_{N2}, X_{ALG}, X_{CON}$   
 Org. material:  $X_S, X_P, S_S, S_I$   
 Nutrients:  $S_{NH4}, S_{NH3}, S_{NO2}, S_{NO3}, S_{HPO4}, S_{H2PO4}$   
 Oxygen:  $S_{O2}$   
 Inorg. material:  $X_P, X_{II}, S_{CO2}, S_{HCO3}, S_{CO3}, S_H, S_{OH}, S_{Ca}$

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## Processes

- Growth & respiration of heterotrophs ( $X_H$ ) (aerobic and anoxic).
- Growth & respiration of nitrifiers ( $X_{N1}, X_{N2}$ ).
- Growth, respiration & death of algae ( $X_{ALG}$ ).
- Growth, respiration & death of consumers ( $X_{CON}$ ).
- Hydrolysis.
- Chemical equilibria.
- Adsorption and desorption of phosphate.

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## Biochemical Conversion Model

Component	$r$	$i_1$	$i_2$	$i_3$	$i_4$	$i_5$	$i_6$	$i_7$	$i_8$	$i_9$	$i_{10}$	$i_{11}$	$i_{12}$	$i_{13}$	$i_{14}$	$i_{15}$	$i_{16}$	$i_{17}$	$i_{18}$	$i_{19}$	$i_{20}$	$i_{21}$	$i_{22}$	$i_{23}$	$i_{24}$	
Process	$r$	$i_1$	$i_2$	$i_3$	$i_4$	$i_5$	$i_6$	$i_7$	$i_8$	$i_9$	$i_{10}$	$i_{11}$	$i_{12}$	$i_{13}$	$i_{14}$	$i_{15}$	$i_{16}$	$i_{17}$	$i_{18}$	$i_{19}$	$i_{20}$	$i_{21}$	$i_{22}$	$i_{23}$	$i_{24}$	
100 Autotrophic Growth of Autotrophs with $NH_4$	1																									
101 Autotrophic Growth of Autotrophs with $NO_3$																										
102 Autotrophic Respiration of $NH_4$																										
103 Autotrophic Respiration of $NO_3$																										
104 Autotrophic Growth of Heterotrophs with $NH_4$																										
105 Autotrophic Growth of Heterotrophs with $NO_3$																										
106 Autotrophic Death of $NH_4$																										
107 Growth of Phosphage Nitrifiers																										
108 Anoxic Respiration of 1st-Stage Nitrifiers																										
109 Anoxic Respiration of 2nd-Stage Nitrifiers																										
110 Anoxic Respiration of 3rd-Stage Nitrifiers																										
111 Growth of Algae with $NH_4$																										
112 Growth of Algae with $NO_3$																										
113 Death of Algae																										
114 Growth of Consumers (no $NH_4$ )																										
115 Growth of Consumers (on $NH_4$ )																										
116 Growth of Consumers (on $NO_3$ )																										
117 Growth of Consumers (on $NH_4$ and $NO_3$ )																										
118 Anabolic Respiration of Consumers																										
119 Death of Consumers																										
120 Hydrolysis																										
121 $SO_4 \leftrightarrow H_2SO_4$																										
122 $CaCO_3 \leftrightarrow Ca^{2+} + CO_3^{2-}$																										
123 $CaCO_3 \leftrightarrow Ca^{2+} + HCO_3^-$																										
124 $CaCO_3 \leftrightarrow Ca^{2+} + CO_2 + H_2O$																										
125 $CaCO_3 \leftrightarrow Ca^{2+} + HCO_3^- + H^+$																										
126 $CaCO_3 \leftrightarrow Ca^{2+} + CO_3^{2-} + H^+$																										
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150 $CaCO_3 \leftrightarrow Ca^{2+} + CO_3^{2-} + H^+$																										

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RIVER WATER QUALITY MODEL NO. 1:  
 III. BIOCHEMICAL SUBMODEL  
 SELECTION

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Contents

$$\frac{\partial c}{\partial t} = -u \frac{\partial c}{\partial x} - v \frac{\partial c}{\partial y} - w \frac{\partial c}{\partial z} + \frac{\partial}{\partial x} \left( \epsilon_x \frac{\partial c}{\partial x} \right) + \frac{\partial}{\partial y} \left( \epsilon_y \frac{\partial c}{\partial y} \right) + \frac{\partial}{\partial z} \left( \epsilon_z \frac{\partial c}{\partial z} \right) + r(c, p)$$

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Contents

$$\frac{\partial c}{\partial t} = -u \frac{\partial c}{\partial x} - v \frac{\partial c}{\partial y} - w \frac{\partial c}{\partial z} + \frac{\partial}{\partial x} \left( \epsilon_x \frac{\partial c}{\partial x} \right) + \frac{\partial}{\partial y} \left( \epsilon_y \frac{\partial c}{\partial y} \right) + \frac{\partial}{\partial z} \left( \epsilon_z \frac{\partial c}{\partial z} \right) + r(c, p)$$

Selection criteria

- ✓ Components
- ✓ Processes
- ✓ General rules for model selection

Examples of model simplifications

- ✓ from  $23 \times 24$  to  $1 \times 2$

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### Utmost simplification: Streeter-Phelps

Component	$i$	(7)	(16)
Process $\downarrow$		$S_{O_2}$	$X_{S_0}$
Aerobic Growth of autotrophs with $NH_4$		-	-
Aerobic Growth of autotrophs with $NO_2$		-	-
Aerobic Growth of autotrophs with $NO_3$		-	-
Aerobic Growth of heterotrophs		-	-
Aerobic Growth of heterotrophs with $NH_4$		-	-
Aerobic Growth of heterotrophs with $NO_2$		-	-
Aerobic Growth of heterotrophs with $NO_3$		-	-
Denitrification		-	-
Denitrification with $NH_4$		-	-
Denitrification with $NO_2$		-	-
Denitrification with $NO_3$		-	-
Aerobic Denitrification		-	-
Aerobic Denitrification with $NH_4$		-	-
Aerobic Denitrification with $NO_2$		-	-
Aerobic Denitrification with $NO_3$		-	-
Anaerobic Growth of heterotrophs		-	-
Anaerobic Growth of heterotrophs with $NH_4$		-	-
Anaerobic Growth of heterotrophs with $NO_2$		-	-
Anaerobic Growth of heterotrophs with $NO_3$		-	-
Anaerobic Denitrification		-	-
Anaerobic Denitrification with $NH_4$		-	-
Anaerobic Denitrification with $NO_2$		-	-
Anaerobic Denitrification with $NO_3$		-	-
Chemotrophic Growth of heterotrophs		-	-
Chemotrophic Growth of heterotrophs with $NH_4$		-	-
Chemotrophic Growth of heterotrophs with $NO_2$		-	-
Chemotrophic Growth of heterotrophs with $NO_3$		-	-
Chemotrophic Growth of autotrophs		-	-
Chemotrophic Growth of autotrophs with $NH_4$		-	-
Chemotrophic Growth of autotrophs with $NO_2$		-	-
Chemotrophic Growth of autotrophs with $NO_3$		-	-
Chemotrophic Growth of autotrophs with $HS^-$		-	-
Chemotrophic Growth of autotrophs with $CO_2$		-	-
Chemotrophic Growth of autotrophs with $CO_3^{2-}$		-	-
Chemotrophic Growth of autotrophs with $CH_4$		-	-
Chemotrophic Growth of autotrophs with $CH_3OH$		-	-
Chemotrophic Growth of autotrophs with $C_2H_5OH$		-	-
Chemotrophic Growth of autotrophs with $CH_3COOH$		-	-
Chemotrophic Growth of autotrophs with $H_2$		-	-
Chemotrophic Growth of autotrophs with $CH_4$		-	-
Chemotrophic Growth of autotrophs with $CO_2$		-	-
Chemotrophic Growth of autotrophs with $CO_3^{2-}$		-	-
Chemotrophic Growth of autotrophs with $CH_4$		-	-
Chemotrophic Growth of autotrophs with $CH_3OH$		-	-
Chemotrophic Growth of autotrophs with $C_2H_5OH$		-	-
Chemotrophic Growth of autotrophs with $CH_3COOH$		-	-
Chemotrophic Growth of autotrophs with $H_2$		-	-

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### Utmost simplification: Streeter-Phelps

Component	$\rightarrow$	$i$	(7)	(16)
$j$	Process $\downarrow$		$S_{O_2}$	$X_{S_0}$
(1+2)	Aerobic Degradation of organic material		-	-

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### Items to consider

- ✓ Relevant compartments ?
- ✓ Variable or constant component conc. ?
- ✓ Nitrite
- ✓ Anoxic conditions
- ✓ Nitrifiers
- ✓ Algae
- ✓ Chemical equilibria
- ✓ General rules

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### Relevant compartments

- ✓ Is it necessary to include (Nr of variables):
  - ✓ Water column
  - ✓ Sediment (pore water and particles)
  - ✓ Biofilm (attached surface)
- ✓ Large river: ~~sediment~~, only water column
- ✓ Small river: large surface - bulk liquid ratio
  - ==> sediment compartment necessary (Case I)
  - ==> biofilm compartment necessary (Case II)

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### State variables or Constants ?

- ✓ Biomass concentrations
  - => algae, consumers, heterotrophs, nitrifiers
  - => especially useful for short term dynamics
- ✓ Component is eliminated from matrix
  - Kinetics are simplified (states ==> constants)
  - Processes are combined (e.g. growth+decay)
- ✓ Note: This does not mean that component concentrations are eliminated from the model !

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### Chemical equilibria

- ✓  $S_{OH} / S_H$       only 2 extra parameters to be estimated !
  - $S_{NH3} / S_{NH4}$
  - $S_{H2CO3} / S_{HCO3} / S_{CO3} / S_{Ca}$
  - $S_{H2PO4} / S_{HPO4}$
- ✓ Normally eliminated
- ✓ Understanding of pH dynamics
- ✓ Possible carbon limitation of algae/nitrif.
- ✓ pH-effects on process rates
- ✓ Ammonia toxicity

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### Simplified model: Example 1

✓ ~~Consumers, pH, P-adsorption/desorption~~

Component	→	j	(1)	(2)	(3)	(5)	(6)	(7)	(8)	(9)	(16)	(17)	(18)	(19)	(21)	(22)
j	Process ↓		S <sub>i</sub>	S <sub>o</sub>	R <sub>bio</sub>	R <sub>aut</sub>	R <sub>NO2</sub>	R <sub>NO3</sub>	S <sub>PO4</sub>	S <sub>O2</sub>	X <sub>o</sub>	X <sub>u</sub>	X <sub>h</sub>	X <sub>s</sub>	X <sub>e</sub>	X <sub>t</sub>
(1a)	Aerobic Growth of Heterotrophs with NH4		-					7	-	-1						
(1b)	Aerobic Growth of Heterotrophs with NO3		-					7	-	-1						
(2)	Aerobic Respiration of Heterotrophs										-1					
(3a)	Anoxic Growth of Heterotrophs with NO2		-					7		-1						
(3b)	Anoxic Growth of Heterotrophs with NO3		-					7		-1						
(4)	Anoxic Respiration of Heterotrophs										-1					
(5)	Growth of 1st-stage Nitrifiers															
(6)	Anoxic Respiration of 1st-stage Nitrifiers															
(7)	Growth of 2nd-stage Nitrifiers															
(8)	Anoxic Respiration of 2nd-stage Nitrifiers															
(9a)	Growth of Algae with NH4															
(9b)	Growth of Algae with NO3															
(10)	Aerobic Regeneration of															
(11)	Death of Algae															
(12)	Hydrolysis															

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### Simplified model: Ex. 2 ≅ QUAL2E

✓ growth is compensated by respiration: X=constant  
hydrolysis is incorporated in degradation rate

Component	→	j	(3)	(5)	(6)	(7)	(9)	(19)	(21)
j	Process ↓		S <sub>NH4</sub>	S <sub>NO2</sub>	S <sub>NO3</sub>	S <sub>HP04</sub>	S <sub>O2</sub>	X <sub>ALC</sub>	X <sub>S</sub>
(1+2)	Aerobic Degradation of organic material		+				+	-	-
(3+4)	Anoxic Degradation of organic material		+		-		+	-	-
(5+6)	Growth and respiration of 1st-stage Nitrifiers		-	+			-		
(7+8)	Growth and respiration of 2nd-stage Nitrifiers		-	+			-		
(9b)	Growth of Algae with NO3		-	-	-	-	+	+	+

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### Simplified model: Example 3 ≅ extended Streeter-Phelps

✓ ~~nitrification + anoxic conditions~~

Component	→	j	(3)	(6)	(7)	(9)	(19)	(21)
j	Process ↓		S <sub>NH4</sub>	S <sub>NO3</sub>	S <sub>HP04</sub>	S <sub>O2</sub>	X <sub>ALC</sub>	X <sub>S</sub>
(1+2)	Aerobic Degradation of organic material		+			+	-	-
(9b)	Growth of Algae with NO3		-	-	-	+	+	+

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Note presence of growing algal biomass !

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## Simplified model: Example 4 ≅ Streeter-Phelps

✓ algae

Component	→	<i>i</i>	(7)	(16)
<i>j</i>	Process ↓		S <sub>O<sub>2</sub></sub>	X <sub>6</sub>
1	(1+2)	Aerobic Degradation of organic material	-	-

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Constant heterotrophic biomass hidden but explicitly present in degradation kinetics!

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## Take home

- ✓ Wake up from your horror dream :
  - ✓ Simplification is nearly always possible
  - ✓ No clear cut decision criteria exist (yet)
  - ✓ Guidelines are available
  - ✓ Some general rules were deduced
  - ✓ Examples were provided (Qual2E, S-P)
- ✓ To be combined with 5 other decision steps

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Ready for applications ?

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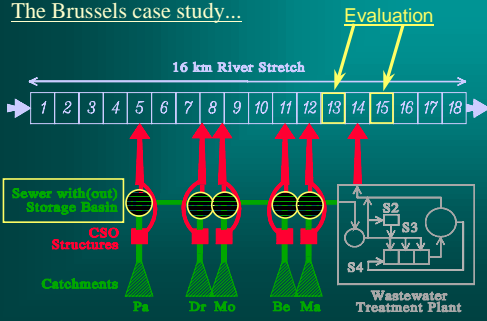
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# Integrated Urban Water Management

The Brussels case study...



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## Effect of 2 design options (BAS/CSO) on River Water Quality



One big and several small rains in summer '86

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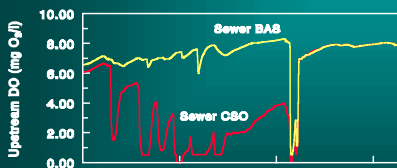
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## River Water Quality (oxygen)

Downstream of CSO, Upstream of WWTP



Clear beneficial effect of retention basins (BAS)!

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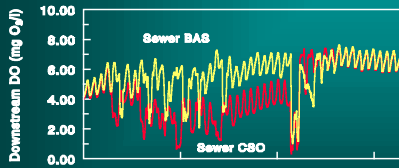
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## River Water Quality (oxygen)

Downstream of CSO, Downstream of WWTP



*Beneficial effect of basins is reduced due to lower efficiency of WWTP by increased loading from basins*

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## GREAT-ER project

Geography-referenced Regional Exposure Assessment Tool for European Rivers

- prediction of the fate of specific “down-the-drain” chemicals in surface water
- using Geographical Information Systems (GIS)
- for use within Environmental Risk Assessment

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## Environmental Risk Assessment

- Aim = assess the probability and severeness of negative effects on ecosystems after exposure to chemicals
- steps:
  - **exposure:** Predicted Environmental Concentration (PEC)  
→ how much ends up in the environment ? where ?
  - **effects:** Predicted No Effects Concentration (PNEC)  
→ how toxic / dangerous is the chemical for the environment ?

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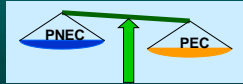
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## Environmental Risk Assessment

- Aim = assess the probability and severeness of negative effects on ecosystems after exposure to chemicals
- steps:
  - **exposure**: Predicted Environmental Concentration (PEC)
  - **effects**: Predicted No Effects Concentration (PNEC)

– risk ratio:  $\frac{PEC}{PNEC} < 1 ?$



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## Environmental exposure assessment

Current methods (advised in EU legislation):  
multimedia fate models:



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## Environmental exposure assessment

- Current methods:
  - multimedia fate models
  - no spatial nor temporal variability considered  
→ limited accuracy

**FACTOR > 100-1000 !**
- GREAT-ER: refine PEC calculations
  - 'real' geo-referenced data
  - variability
  - geo-referenced → validation is possible

**AIM = FACTOR < 3-5**

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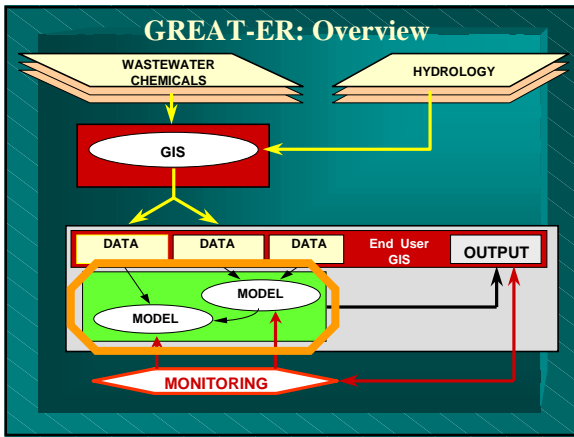
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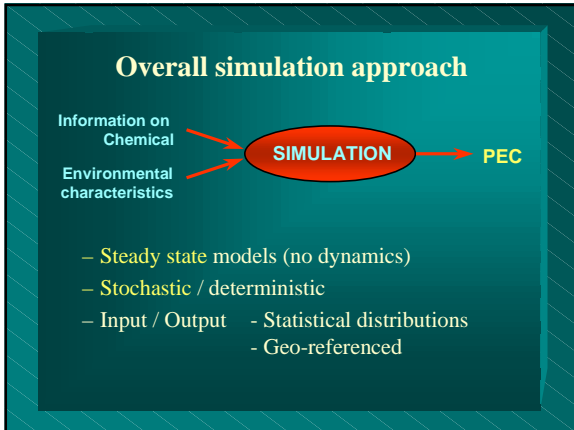
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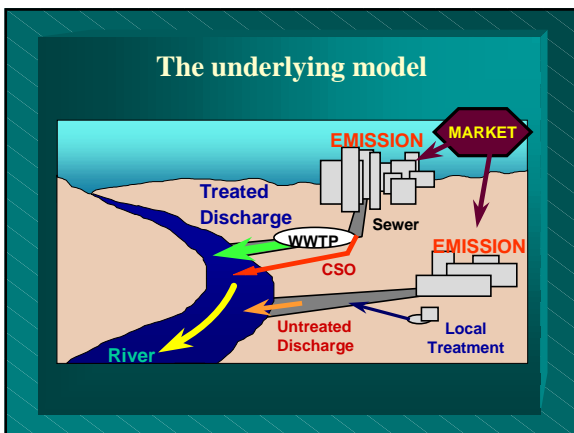
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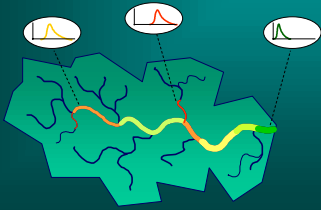
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### Simulation Data Analysis

- direct GREAT-ER simulation results: geo-referenced predicted concentrations



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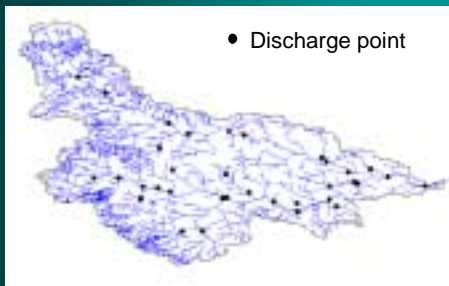
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### GREAT-ER Validation

Catchment in Yorkshire (UK)



- Discharge point

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### GREAT-ER Validation

LAS concentrations in the river stretches



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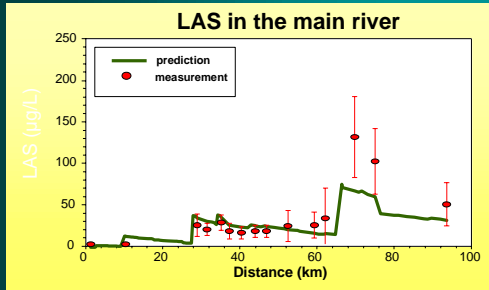
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## Validation of GREAT-ER



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## Case study Integrated Model: Lambro

Treatment plant of Merone (north of Milano, Italy)

- overloaded during day time under dry conditions
- expansion of plant with increased hydraulic capacity

Effect of upgrade on river water quality is assessed using an integrated model:

- WWTP (anoxic/aerobic activated sludge)
- River (modified RWQM1, 47 tanks)

with WEST modelling & simulation platform

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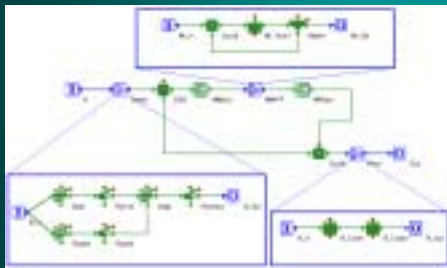
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## Case study Integrated Model: Lambro Implementation in WEST (hierarchical !)



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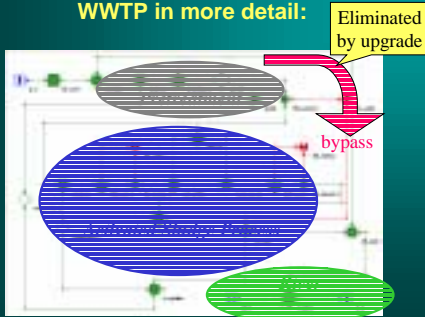
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## Case study Integrated Model: Lambro WWTP in more detail:



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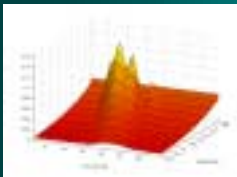
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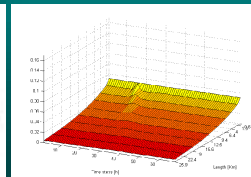
## Case study Integrated Model: Lambro

The LAS concentration is followed along the river as function of time and distance

Before plant upgrade



After plant upgrade



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## Conclusions

**INTEGRATED WATER QUALITY MODELLING  
HAS BEEN MUCH FACILITATED THANKS TO:**

- ☞ New RWQMI conversion model that is ASM-compatible
- ☞ Suggested guidelines for (simple) submodel selection
- ☞ Increased interest in integrated analysis by authorities leads to an increasing number of illustrative case studies
- ☞ Open modelling & simulation platform (WEST)

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