



Equilibrium temperature in aerated basins—comparison of two prediction models

Sylvie Gillot^{a,*}, Peter A. Vanrolleghem^b

^a Cemagref, Parc de Tourvoie—BP 44, F-92163 Antony cedex, France

^b BIOMATH, Ghent University, Coupure Links 653, B-9000 Ghent, Belgium

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Abstract

This note presents and compares two models to predict the equilibrium temperature in aerated basins. They differ by their degree of complexity and therefore by the input data they require. Both models were able to estimate the temperature of an industrial aerated lagoon, the more complex model giving, in addition, a complete breakdown of the heat exchanges.

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1. Introduction

In biological wastewater treatment processes, operating temperature is highly important as it significantly influences treatment performance, e.g. by increasing pollutant conversion rates typically by a factor when the temperature increases with 10 degrees. Several models have been developed to predict this process temperature: (i) during the design phase of a project [1,2] or (ii) during operation [3–5]. Even dynamic temperature changes in tanks have been successfully modelled [6–8]. These models incorporate the different heat gains/losses over the basins. They however require a large number of input data (meteorological and operating conditions), which are not always easy to collect especially at the design stage of a plant.

On the other hand, few authors developed more simple formulae [9,10]. Among them, van der Graaf [11] proposed a simple, semi-empirical model to evaluate aeration tank temperature, which requires only little data.

The objective of this technical note is to compare the simple model proposed by van der Graaf [11] with a rather complete model developed on the basis of the work by Talati and Stenstrom (1991). Both models are described and they are compared while estimating the equilibrium temperature of an aerated lagoon treating industrial wastewater in the North of France. A comparison of the heat contributions of different processes is made as well.

2. Material and methods

2.1. Heat balance

The equations presented herein have been developed for a completely mixed basin under steady state conditions. The completely mixed hypothesis supposes that the water temperature is uniform over the basin, and equals the outlet temperature. The energy balance over the reactor implies that the net heat exchange (Q_t) equals the enthalpy change between the influent and the effluent streams (ΔH). The latter is written as

$$\Delta H + \rho_w c_{pw} Q_w (T_i - T_w) = 0, \quad (1)$$

*Corresponding author. Tel.: +33-1-40-96-60-66; fax: +33-1-40-96-61-99.

E-mail address: sylvie.gillot@cemagref.fr (S. Gillot).

where ρ_w is the density of water, kg/m^3 ; c_{pw} the specific heat of water, J/kg/K ; Q_w the wastewater flow rate, m^3/s ; T_i the influent temperature, K ; and T_w the water temperature, K .

The two models presented differ in the terms they include to calculate the net heat exchange.

2.1.1. Simple model: van der Graaf [11]

Fig. 1 gives an overview of the heat exchange terms considered in the van der Graaf’s model.

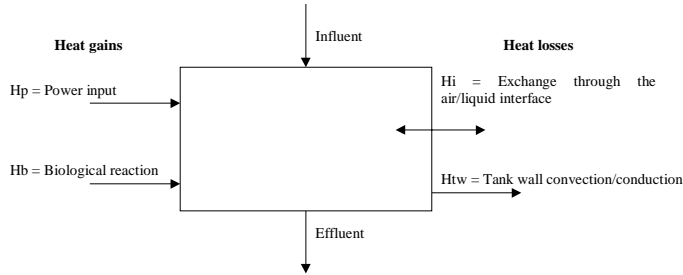


Fig. 1. Overview of the heat exchanges over the basin according to the simple model.

Table 1
Heat losses/gains—Simple model

Term	Equation	Parameter		Value	Unit	
		Symbol	Description			
Hi	$U_i A(T_a - T_w)$	U_i	Heat coefficient	Calculated	$\text{W/m}^2 \text{K}$	
		A	Basin area	Variable	m^2	
		Subsurface aeration: $U_i = 25$	T_a	Air temperature	Variable	K
		Surface aeration: $U_i = 11.4NP_{\text{aer}}/V$	T_w	Water temperature	Variable	K
			V	Volume of the tank	Variable	m^3
			P_{aer}	Aerator power	Variable	W
Htw	$H_{tw} = U_w A_g(T_w - T_c)$	U_w	Overall heat transfer coefficient for basin wall/bottom	Variable	$\text{W/m}^2 \text{K}$	
		T_c	Soil temperature	Variable	K	
		A_g	Wall + bottom area	Variable	m^2	
		P_{aer}	Aerator power	Variable	W	
Hp	$H_p = P_{\text{aer}} N$	N	Number of aerators			
Hb	$H_b = 4.1 \text{OC}(C_s - C)/C_s$	OC	Oxygenation capacity	Variable	kgO_2/h	
		C_s	Concentration of oxygen at saturation	Calculated	mg/L	
		C	Concentration of oxygen in the tank	Variable	mg/L	

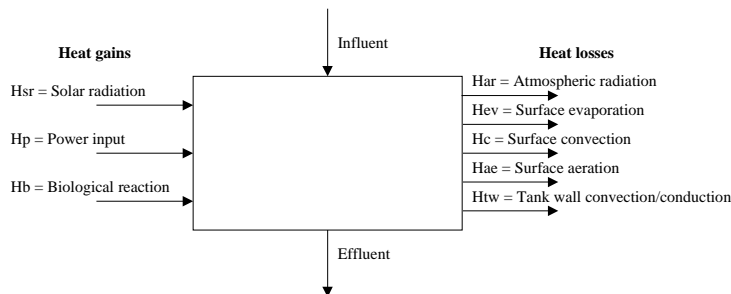


Fig. 2. Overview of the heat exchanges over the basin according to the complete model.

The net heat exchange equals (see Fig. 1 for the separate terms and Table 1 for the underlying relationships)

$$\Delta H = H_p + H_b - H_i - H_{tw} \tag{2}$$

2.1.2. Complete model: Talati and Stenstrom [5]

Fig. 2 gives an overview of the heat exchange terms considered in the complete model.

The net heat exchange equals (see Fig. 2 for the separate terms and Table 2 for the underlying

Table 2
Heat losses/gains—Complete model

Term	Equation	Parameter		
		Symbol	Description	Unit
Har	$\varepsilon\sigma(T_w + 273)^4 A - (1 - \lambda)\beta\sigma(T_a + 273)^4 A$	ε	Water emissivity	Dimensionless
		σ	Stefan Boltzman constant	$W/m^2/K^4$
		T_w	Water temperature	Variable
		A	Tank area	Variable
		λ	Water reflectivity	0.03
		β	Atmospheric radiation factor	Variable
		T_a	Air temperature	Variable
Hev	$Hev = 4.18/3600/24 \times (1.145 \times 10^6(1 - r_h/100) + 6.86 \times 10^4(T_w - T_a))e^{0.0604T_a} WA^{0.95}$	r_h	Relative humidity	%
Hc	$Hc = \rho_a c_{pa} h_v A (T_w - T_a)$	W	Wind speed	m/s
		h_v	Vapour phase transfer coefficient	Calculated
Hae	$h_c = 392A^{-0.05} W/3600 \times 24$ $\rho_a = 1.293(273/(273 + T_a))$ $Ha = Has + Hal$ Surface aeration: $Has = h_v \rho_a c_{pa} (T_w - T_a) A$ $h_v = 392F^{-0.05} W/3600 \times 24$ Subsurface aeration: $Has = Q_a \rho_a c_{pa} (T_w - T_a) A$ Surface aeration: $Hal = M_w Q_a L_s / 100 / R((v_w(r_h + h_f)(100 - r_h))/(T_w + 273)) - v_a r_h / (T_a + 273)$	ρ_a	Density of air	kg/m ³
		c_{pa}	Specific heat of air	J/kg/K
		Has	Sensible heat loss	1050
		Hal	Evaporative heat loss	Calculated
		H_v	Vapour phase transfer coefficient	Calculated
		W	Wind speed	Variable
		F	Spray area of the aerators	Variable
Qa	$Qa = NFW$	Q_a	Air flow rate	m ³ /s
		M_w	Molecular weight of water	g/mol
		L_s	Latent heat of evaporation	2270
		R	Universal gas constant	62.631
		N	Number of aerators	Variable
v_w	v_w	v_w	Vapour pressure at water temp.	mmHg
		Y_a	Vapour pressure at air temp.	mmHg
		h_f	Exit air humidity factor	Variable
		N	Number of aerators	Variable

H _{tw}	$H_{tw} = U_w Ag(T_w - T_e)$	U_w	Overall heat transfer coefficient for basin wall/bottom	Variable	W/m ² K
H _{sr}	$H_{sr} = H_{sr,0}(1 - 0.0071 C_e^2)A$ $H_{sr,0} = (a - b \sin(2\pi d/366 + c))4.18$ $a = 95.1892 - 0.3591k - 8.4537 \times 10^{-3}k^2$ $b = -6.2484 + 1.6645k - 1.1648 \times 10^{-2}k^2$ $c = 1.4451 + 1.434 \times 10^{-2}k - 1.745 \times 10^{-4}k^2$ $H_p = NP_{aer}$ $H_b = h\Delta S$	T_e Ag $H_{sr,0}$ K C_e d P_{aer} h ΔS	Soil temperature Wall + bottom area Solar radiation for clear sky conditions Latitude of the site Cloud cover Day of the year (1–365) Aerator power Heat produced from degradation of organics Substrate removal rate	Variable Variable Variable Variable Variable Variable 7531 Variable	K m ² W/m ² Degree Tenths Dimensionless W J/gCOD gCOD/s

relationships)

$$\Delta H = H_{sr} + H_p + H_b - H_{ar} - H_{ev} - H_c - H_{ae} - H_{tw} \tag{3}$$

A complete description of the model can be found in Talati and Stenstrom [5] and Sedory and Stenstrom [8].

As seen in Table 2, the complete model requires an important number of input data, with different levels of uncertainty. To reduce this number, Namèche and Vassel [4] proposed to measure the solar radiation and the evaporation, rather than predict them with the model.

The heat balance over the basin consists in finding the water temperature for which the net heat exchange equals the enthalpy exchange (Eq. (1)). The solver of Excel has been used for this purpose.

2.2. Application to an aerated lagoon

The models have been applied to an aerated lagoon (10,000 m³) located in the North of France, treating industrial wastewater (10–40 tonnes of COD per day). During winter time, the lagoon’s COD removal efficiency is lower due to low temperature. Data were either measured at the plant or provided by a local weather station. A 10-day measuring campaign performed in April 2000 was first used to validate the models. Estimated temperatures throughout the year 2000 for each and every month were further calculated to compare the models between them.

The data required to estimate the equilibrium temperature using the model of van der Graaf [11] or the model of Talati and Stenstrom [5] are summarized in Tables 3 and 4 for the estimation concerning the 10-day measuring campaign. Table 5 presents the data used to estimate the monthly average temperature.

3. Results and discussion

3.1. Heat exchange terms

The heat exchanges considered in both models are depicted in Figs. 3 and 4.

In both models the heat losses through the walls can be neglected. While the simplified model aggregates all the terms involving a heat transfer through the gas/liquid interface ($H_i = H_{sr} + H_{ae} + H_c + H_{ev} + H_{ar}$), the complete one details these terms.

3.2. Temperature prediction

The temperatures predicted by both models have been compared to the measured temperature during the 10-day measuring campaign, and are presented in Table 6.

Table 3
Data required in the simple model

Parameter	Value	Unit	Source
Tank area, A	4500	m ²	Design data
Tank volume, V	10,000	m ³	Design data
Wastewater flow rate, Q_w	0.00525	m ³ /s	On site measurement
Influent temperature, T_i	23.3	°C	On site measurement
Power input, P	55,000	W	Design data
Heat transfer coeff. wall, U_w	1	W/m ² /°C	Estimated
Air temperature, T_a	8.2	°C	On site measurement
Soil temperature, T_e	8	°C	Estimated
Oxygenation capacity, OC	16	kg O ₂ /h	Design data
Oxygen concentration at saturation, C_s	11.3	mg/L	Estimated
Oxygen concentration, C	2	mg/L	On site measurement
Wall + bottom area A_g	4500	m ²	Design data

Table 4
Data required in the complete model

Site specific data			Meteorological data ²			Process parameter ³		
Parameter	Value	Unit	Parameter	Value	Unit	Parameter	Value	Unit
Latitude, k^a	50	°	Vapour pressure at, $T_w v_w^b$	8.9	mmHg	Wastewater flow rate, Q_w^c	0.00525	m ³ /s
Tank area, A^a	4500	m ²	Vapour pressure at, $T_a v_a^b$	6.4	mmHg	Influent temperature, T_w^c	23.3	°C
Number of aerators, N^a	5	—	Air temperature, T_a^d	8.2	°C	Substrate removal rate, ΔS^c	0.021	kgCOD/s
Aerator spray area, F^a	6	m ²	Wind speed, W^d	5.0	m/s	Day of the year d	106	
Power input, P^a	55	kW	Relative humidity, r_h^d	79	—			
Air Humidity factor, h_f^b	0.9		Cloud cover, C_c^d	5	Tenths			
Heat transfer coeff. wall, U_w^b	1	W/m ² /°C	Atmospheric radiation factor, β^b	0.8				
			Soil temperature, T_e^b	8	°C			

^a Design data.

^b Estimated.

^c Daily measurements.

^d From a meteorological station located in Lille, 30 km from the wastewater treatment plant.

Table 5
Data used to perform monthly estimation throughout the year 2000

Month	J	F	M	A	M	J	J	A	S	O	N	D
T_a (°C) ¹	3.0	3.0	6.0	8.0	12.0	15.0	17.0	17.0	15.0	11.0	6.0	4.0
W (m/s) ¹	6.1	6.7	6.7	5.0	4.4	4.4	4.4	3.9	4.4	4.4	5.0	5.6
r_h^a	88	85	82	79	78	79	78	78	83	87	89	90
Q_w (m ³ /s) ^b	0.26	0.26	0.27	0.27	0.29	0.29	0.31	0.31	0.33	0.29	0.26	0.23
T_{in} (°C) ²	25.8	25.1	27.8	28.5	27.5	28.0	31.0	31.3	29.0	29.5	27.5	21.7
ΔS (tCOD/d) ^b	27.7	18.6	19.5	20.1	21.1	21.1	31.8	30.4	38.1	36.4	27.9	12.7

^a From the meteorological station.

^b Monthly averages from daily measurements at the plant.

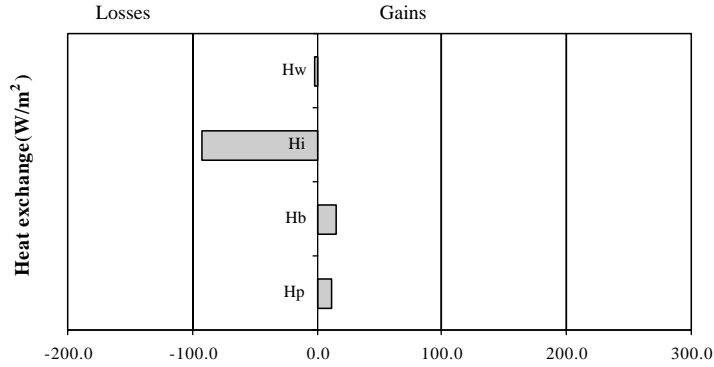


Fig. 3. Heat exchanges according to the simple model.

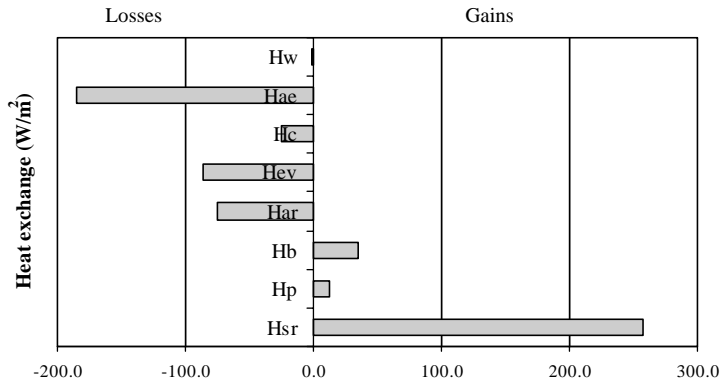


Fig. 4. Heat exchanges according to the complete model.

Table 6
Equilibrium temperatures estimated by the models

T_{measured} (°C)	Simple model			Complete model		
	$T_{\text{estimated}}$ (°C)	Diff (°C)	Relat. Diff (%)	$T_{\text{estimated}}$ (°C)	Diff (°C)	Relat. Diff. (%)
10.1	9.7	-0.4	-4.1	10.0	-0.1	-0.7

Both models give a good estimation of the equilibrium temperature.

3.3. Monthly averages prediction

The models have been used to predict the monthly average temperatures in the lagoon obtained on the basis of meteorological and operational data for the year 2000. The estimated temperatures are presented in Fig. 5. The temperatures predicted by both models are close. They may therefore both be used to estimate the temperature variation through the year.

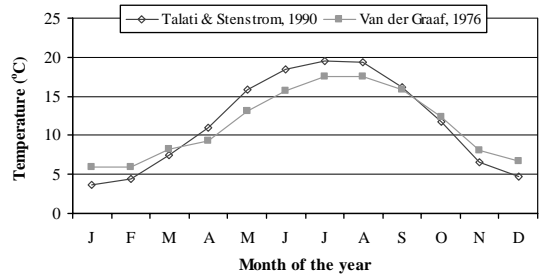


Fig. 5. Prediction of monthly average temperature (T_w).

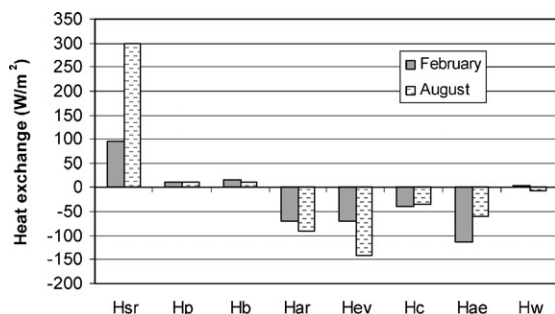


Fig. 6. Comparison of heat exchange breakdowns in winter and summer.

3.4. Breakdown of the heat exchanges

Using the complete model, the heat exchange terms per m^2 of tank area have been calculated for February and August. The obtained breakdowns are depicted in Fig. 6.

The results obtained show that the solar radiation is the main energy source for the lagoon (75% and 95% of the heat gain in winter and summer, respectively). Energy is mainly lost through aeration, and evaporation and radiation (20–40% of the heat loss for each). These results are similar to the literature data [5,4]. Fig. 6 also points out that if the energy gains/losses are rather well-balanced during warm weather, the losses are more important during winter, which may induce a significant decrease in temperature and the consequent COD removal deficiencies encountered.

The fact that the estimated temperature with the complete model is higher in summer and lower in winter than the one calculated with the simple model (see Fig. 5) may be attributed to the difference observed in the solar radiation, much higher in summer time. This difference is not directly taken into account in the simple model.

4. Conclusions

This note described two models to estimate the equilibrium temperature in aerated basins. They differ by their degree of complexity and therefore by the input data they require. The equations used to calculate the different heat exchange terms are presented, and the models are applied to data of an aerated lagoon. Both models have been validated using a 10-day measuring campaign: they give a temperature close to the measured

one. They also give similar monthly estimated temperature, on the basis of meteorological data and process performances. Therefore they may both be used in this case to estimate the equilibrium temperature, the more complex model giving, in addition, a complete breakdown of the heat exchanges.

The models should however be further validated on data obtained for basins under different conditions (e.g. other types of aeration systems, such as horizontal rotors, diffused aeration, sites with different meteorological conditions, nitrifying plants, etc.).

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References

- [1] Brown EV, Enzinger JD. Temperature profile and heat transfer model for a chemical wastewater treatment plant. *Environ Prog* 1991;10:159–68.
- [2] La Cour Jansen J, Kristensen GH, Laursen KD. Activated sludge nitrification in temperate climate. *Water Sci Technol* 1992;25:177–84.
- [3] Argaman Y, Adams CE. Comprehensive temperature model for aerated biological systems. *Prog Water Technol* 1977;9:397–409.
- [4] Namèche T, Vassel J-L. Bilan thermique sous climat tempéré des lagunes aérées et naturelles. *Rev Sci l'Eau* 1999;12:65–91.
- [5] Talati SN, Stenstrom MK. Aeration basin heat loss. *J Environ Eng* 1990;116:70–86.
- [6] Makinia J, Wells SA. A general model of the activated sludge reactor with dispersive flow—I. Model development and parameter estimation. *Water Res* 2000;34:3987–96.
- [7] Scherfig J, Schleisner L, Brond S, Klide N. Dynamic temperature changes in wastewater treatment plants. *Water Environ Res* 1996;68:143–51.
- [8] Sedory PE, Stenstrom MK. Dynamic prediction of wastewater aeration basin temperature. *J Environ Eng* 1995;121:609–18.
- [9] Eckenfelder WW. *Industrial water pollution control*, 2nd edition. New York: McGraw-Hill; 1989.
- [10] Grady CPL, Daigger GT, Lim HC. *Biological wastewater treatment*. 2nd edition, Revised and Expanded. New York: Marcel Dekker, 1999.
- [11] van der Graaf JHJM. Laten biologische zuiveringsprocessen zich naar temperatuur optimaliseren. *H₂O* 1976;9:87–93 (in Dutch).